November 2011 Synthesis of Novel 1,7-Naphthyridines by Friedländer 1383 Condensation of Pyridine Substrates

Vegar Stockmann and Anne Fiksdahl*

Department of Chemistry, Norwegian University of Science and Technology, NTNU, N-7491 Trondheim, Norway *E-mail: Anne.Fiksdahl@chem.ntnu.no Received June 15, 2010 DOI 10.1002/jhet.657 Published online 3 August 2011 in Wiley Online Library (wileyonlinelibrary.com).

The general ability of appropriate pyridyl compounds (aldehyde or ketone) to undergo Friedländer condensation to give different 1,7-naphthyridines has been demonstrated. 2,4-Disubstituted 1,7-naphthyridine 8 was prepared from 3-amino-4-acetylpyridine (6) and ketone 4 (82%) . The Friedlander self-condensation of pyridyl substrate 6 is reported, as well. The dimer product, 2-(3-aminopyridin-4-yl)-4 methyl-1,7-naphthyridine (7), was obtained in 97% yield. 2-Aryl- and 2,3-diaryl-1,7-naphthyridines (16– 18) were prepared from 3-aminoisonicotinaldehyde (13) and arylketones 4, 14, and 15 (28–71%). The key substrates 6 and 13 are readily available via the improved pyridine nitration method.

J. Heterocyclic Chem., 48, 1383 (2011).

INTRODUCTION

The Friedländer reaction is a cyclisation method consisting of (i) a base- or acid-promoted aldol condensation of an aromatic 2-amino-substituted carbonyl compound 1, with an appropriate ketone 2, possessing a reactive α -CH₂ group and (ii) an amine-carbonyl cyclodehydration to form an imine moiety (Scheme 1). The Friedländer annulations are often carried out by refluxing an ethanolic solution of the reactants in the presence of NaOH. This cyclocondensation method is widely used in heterocyclic chemistry, in particular, for the preparation of substituted quinolines 3. Recent advances in the Friedländer reaction [1], as well as the Friedländer approach for quinoline synthesis [2] have been reviewed. New protocols for the preparation of quinoline derivatives by Friedländer annulation reactions have lately been reported [3–7].

The classical Friedländer reaction conditions make use of 2-aminobenzaldehyde $(1, R=H)$ to afford quinolines, I, 3 (Scheme 1). However, in recent years, the substrate has been replaced with 2-amino-nicotinaldehyde, allowing the preparation of some 1,8-naphthyridines, II [5,8–14]. Naphthyridines provide an important scaffold for a variety of compounds of unique biological activities. Their synthesis, properties, reactivity and biological activity are covered in several reviews [15–17]. The synthetic use of the Friedländer reaction for preparation of 1,7-naphthyridines (III) has, however, been limited by the inconvenient preparation methods for the necessary 2-amino carbonyl compounds, such as 3 aminoisonicotinaldehyde (13) [18]. Therefore, the Friedländer approach has nearly not been applied for the preparation of 1,7-naphthyridines [19] and the biological activity of 1,7-naphthyridines has been less studied. 1,7- Naphthyridines are, however, reported to be more active than the corresponding 1,8-isomers as potential new therapeutic antitumor agents [20]. Recently, 1,7-naphthyridine derivatives have been identified as selective Tpl2 kinase inhibitors. Tpl2 is an attractive target for the treatment of rheumatoid arthritis [21].

Based on the fact that a number of substituted 3-nitropyridines have become readily available through an improved nitration method [22,23], we have access to appropriate o-amino-4-carbonylpyridine substrates, such as 3-amino-4-acetylpyridine (6, Scheme 2) and 3-amino-4-pyridinecarboxaldehyde (13, Scheme 3), for the preparation of 1,7-naphthyridines. The present results on 1,7 naphthyridine syntheses are part of an investigation of

the chemistry of nitropyridines, which is in progress in our laboratories.

RESULTS AND DISCUSSION

3-Amino-4-acetylpyridine (6) was prepared by nitration and reduction (Scheme 2) [24]. The dual functionality of aminopyridylketone 6 enables an internal Friedländer self-condensation. In fact, substrate 6 underwent a Friedländer dimerisation reaction by treatment of excess (1.5 equiv) NaH in dry THF at $0-20^{\circ}$ C in 2 h. The 1,7-naphthyridine product 7 was obtained in quantitative yield, which is exceptional compared to the highest yields (80–90%) normally reported for the Friedländer reaction [1]. The structure was unambiguously confirmed by HMBC and HSQC experiments. NOESY experiments of product 7 showed a through-space proximity between H_3 and both C_4 -CH₃ and pyridine-H₅ as well as between C_4 -CH₃ and H₅.

Aminoketones are less frequently used in the Friedländer reactions than aminoaldehydes, since the selfcondensation, as discussed above, may represent a problem and therefore limit the scope and generality of the reaction. Therefore, the capability of 3-amino-4-acetylpyridine (6) to undergo regular Friedländer condensation with other ketones and, thus, to exclude dimerisation, was studied. Indeed, treatment of methylpyridylketone 4 with NaH in THF and subsequent addition of substrate 6, afforded the 2,4-disubstituted 1,7-naphthyridine 8. The ratio between the desired product 8 and the dimer product 7 increased by reducing the amount of NaH from 2.5 to 1.1 equivalent. Rising the initial reaction temperature from 0 to 20° C to assure full deprotonation of ketone 4 before the addition of substrate 6, increased the yield of the target product 8 twofold. As a result, the optimized reaction conditions allowed the isolation of 1,7-naphthyridine 8 in 82% yield.

3-Aminoisonicotinaldehyde (13) was also used as a substrate to demonstrate the general potential of pyridyl compounds for the preparation of 1,7-naphthyridines by the Friedländer condensation (Scheme 3). Methyl 3-aminoisonicotinate (11), readily accessible from methyl isonicotinate (9) by nitration and reduction [24], was transformed into the corresponding Weinreb amide (12, 73%), using the Me₂AlCl/MeONHMe·HCl reagent system $[25]$. Following reduction with LiAlH₄ afforded 3aminoisonicotinaldehyde (13, 90%).

Friedländer reactions of pyridyl substrate 13 with the respective ketones 4, 14, and 15 and 2–2.5 equiv NaH in THF gave the 2-aryl and 2,3-diaryl-1,7-naphthyridines 16–18. Ketone 15 was prepared from isonicotinaldehyde via 4-dimethoxymethylpyridine, BuLi proton abstraction, subsequent nucleophilic substitution of benzylchloride and hydrolysis [26]. Ketones 4 and 14 were commercially available. Product 16 has previously been prepared by standard Friedländer reaction conditions (NaOH/EtOH) in lower yield (60%) [19] than obtained by our alternative NaH/THF method (71%). The novel products 17 and 18 were obtained in 28–31% yield, due to the formation of unidentified by-products. The yields of products 16–18 were not influenced by the order of reactant addition. HMBC and HSQC data obtained by

Reagents and conditions: (i) 1. N_2O_5 in MeNO₂, 0 °C, 2. NaHSO₃ in MeOH/H₂O; (ii) Na₂S₂O₄ in EtOH, reflux, 6 h; (iii) NaH, THF 0 $^{\circ}$ C \rightarrow rt, 2 h; (iv) NaH, THF, rt, 2 h.

November 2011 Synthesis of Novel 1,7-Naphthyridines by Friedländer 1385 Condensation of Pyridine Substrates

Reagents and conditions: (i) 1. N_2O_5 in MeNO₂, 0 °C, 2. NaHSO₃ in MeOH/H₂O; (ii) Na₂S₂O₄ in EtOH, reflux, 6 h; (iii) MeONHMe·HCl/Me₂AlCl in DCM, rt, 20 h; (iv) LiAlH₄ in THF, -15 °C; (v) NaH in THF, $0^{\circ}C$ - rt, 2 h.

2D NMR experiments of products 16–18 confirmed the respective structures.

CONCLUSION

Optimized reaction conditions allowed Friedländer cyclocondensation of substrate 6 and methylpyridylketone 4 to afford 2,4-disubstituted 1,7-naphthyridine 8 (82%). 3-Amino-4-acetylpyridine (6) underwent a Friedländer self-condensation by treatment of excess NaH to afford the 1,7-naphthyridine product 7 in quantitative yield. 3-Aminoisonicotinaldehyde (13) reacted by Friedländer condensation with arylketones 4, 14, and 15 to give 2-aryl- and 2,3-diaryl-1,7-naphthyridines 16–18 (28–71%). Substrates 6 and 13 were readily obtained by pyridine nitration and reduction. Thus, the present results demonstrate that the pyridine nitration pathway followed by Friedländer condensation represent a convenient strategy for the preparation of 1,7 naphthyridines.

EXPERIMENTAL

General. Chemicals: NaH, LiAlH₄, MeONHMe-HCl, Me2AlCl (Aldrich), 4-acetylpyridine (4, Fluka), 1-acetonaphthone (14, Aldrich). 3-Amino-4-acetylpyridine (6) [24], methyl 3-aminoisonicotinate (11) [24], and 2-phenyl-1-(pyridine-4yl)ethanone (15) [26] were prepared as described in literature. Solvents: pro analysi quality. Dry THF and DCM were collected from a MB SPS-800 solvent purification system. All reactions were performed under argon atmosphere in predried glassware. NMR: Bruker Avance DPX 400 MHz. ¹H and ¹³C chemical shifts are reported in ppm downfield from TMS. J values are given in Hz. ESI-MS accurate mass determination was performed on a Waters QTOF II instrument. IR: Nicolet 20SXC FT-IR spectrophotometer. IR spectra were recorded using a Smart Endurance reflexion cell, unless KBr are specified. All melting points are uncorrected and were recorded on a Stuart apparatus. Flash chromatography: $SiO₂$ (SDS, 60 Å, $40-63 \mu m$).

2-(3-Aminopyridin-4-yl)-4-methyl-1,7-naphthyridine (7). A solution of amine 6 (113 mg, 0.830 mmol) in dry THF (3 mL) was added dropwise over 10 min to a solution of NaH (30.0 mg, 1.25 mmol) in dry THF (2 mL) at 0°C and kept stirring for 15 min. The reaction was allowed to heat to room temperature and stirred for 2 h before quenching with water (15 mL). The mixture was extracted with EtOAc $(4 \times 20 \text{ mL})$ and the combined organic phases were dried over Na₂SO₄. After evaporation of the solvent, the crude product was purified by flash column chromatography (gradient; $5-10\%$ MeOH/CH₂Cl₂) to give 95 mg (97%) of the title compound 7 as a yellow solid, mp 250–251°C, pure by NMR; R_f 0.35 (10% MeOH/CH₂Cl₂); IR (KBr): 3348, 3244, 3123, 1602, 1561, 1503, 1434, 1240, 1215 cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ_H 9.47 (d, $J = 0.8$ Hz, 1H, H8), 8.67 (d, $J = 5.6$ Hz, 1H, H6), 8.27 (s, 1H, py'-H2), 8.07 (d, $J = 5.2$ Hz, 1H, py'-H6), 7.93 (s, 1H, H3), 7.80 (dd, $J = 5.6$, 0.8 Hz, 1H, H5), 7.55 (d, $J = 5.2$ Hz, 1H, py'-H5), 6.31 (br s, 2H, -NH2), 2.78 (s, 3H, C4-CH3); 13C NMR

(100 MHz, CDCl₃): δ_C 158.2 (C2), 153.9 (C8), 144.8 (C4), 144.2 (C6), 142.9 (py'-C3), 141.7 (C8a), 140.9 (py'-C2), 138.3 (py'-C6), 130.2 (C4a), 124.9 (py'-C4), 123.6 (C3), 121.9 (py'-C5), 116.3 (C5), 18.5 (C4-CH3); NMR assignments are based on HMBC, HSQC, and NOESY experiments; ESI-HRMS: calcd for $[M + H]^+$ C₁₄H₁₃N₄: 237.1135; obsd 237.1151; calcd for $[M + Na]^+$ C₁₄H₁₂N₄Na: 259.0954; obsd 259.0962.

4-Methyl-2-(pyridin-4-yl)-1,7-naphthyridine (8). To a solution of NaH (18.0 mg, 0.750 mmol) in dry THF (2 mL) at 0° C was added ketone 4 (100 mg, 0.825 mmol). The reaction was stirred for 15 min at 0° C and then 15 min at room temperature. Amine 6 (94.0 mg, 0.690 mmol) in THF (1 mL) was added dropwise over 15 min and the reaction was stirred for 2 h at room temperature. The reaction mixture was diluted with THF (1 mL) before it was added to water (20 mL). After extraction with EtOAc (3×20 mL), drying over NaSO₄ and concentration under reduced pressure, the crude product was purified by flash chromatography (gradient: 5–10% MeOH/ CH_2Cl_2) to give 126 mg (82%) of the title compound 8 as a white solid, mp $135-136^{\circ}$ C, pure by NMR. Compound 7 was isolated as a by-product (8 mg, 10%). 8: R_f 0.15 (5% MeOH/ CH₂Cl₂); IR: 1595, 1417, 837, 829, 750, 685 cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ_H 9.57 (s, 1H, H8), 8.81 (d, $J = 5.6$ Hz, 2H, py'-H2, -H6), 8.67 (d, $J = 5.6$ Hz, 1H, H6), 8.06 (d, $J =$ 5.6 Hz, 2H, py'-H3, -H5), 7.92 (s, 1H, H3), 7.79 (d, $J = 5.6$ Hz, 1H, H5), 2.79 (s, 3H, C4-CH₃); ¹³C NMR (100 MHz, CDCl₃): δ_C 155.9 (C2), 155.4 (C8), 150.8 (py'-C2, -C6), 146.0 (py'-C4), 145.2 (C4), 144.3 (C6), 143.2 (C8a), 131.2 (C4a), 122.7 (C3), 121.7 (py'-C3, -C5), 116.4 (C5), 18.6 (C4-CH3); NMR assignments are based on HMBC and HSQC experiments; ESI-HRMS: calcd for $[M + H]^{+}$ C₁₄H₁₂N₃: 222.1026; obsd 222.1029.

3-Amino-N-methoxy-N-methylisonicotinamide (12). To a solution of MeONHMe-HCl $(2.00 \text{ g}, 20.5 \text{ mmol})$ in dry CH_2Cl_2 (50 mL) at 0° C was added Me₂AlCl (1 *M* in hexanes, 20.5 mL, 20.5 mmol) dropwise over 30 min. The reaction was allowed to heat to room temperature over 2 h. A solution of amine 11 $(1.265 \text{ g}, 8.21 \text{ mmol})$ in dry CH_2Cl_2 (50 mL) was added and the reaction was stirred for 20 h. A solution of borate buffer (pH 8.0, 80 mL) was added and stirring was continued for 10 min. Extraction with CH₂Cl₂ (4 \times 60 mL), drying over Na₂SO₄, concentration under reduced pressure and flash chromatography (gradient: $5-10\%$ MeOH/CH₂Cl₂) afforded the title compound 12 as a white solid, 1.09 g (73%) , mp 94–95°C, pure by NMR; R_f 0.17 (5% MeOH/CH₂Cl₂); IR: 3449, 3310, 3141, 1623, 1583, 1418 , 1384, 983, 968, 832 cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ_H 8.18 (s, 1H, py-H2), 7.98 (d, $J = 4.8$ Hz, 1H, py-H6), 7.25 $(d, J = 4.8 \text{ Hz}, 1H, \text{ py-H5}), 4.64 \text{ (br s, 2H, NH}_2), 3.57 \text{ (s, 3H)}$ -OCH₃), 3.37 (s, 3H, -CH₃); ¹³C NMR (100 MHz, CDCl₃): δ_C 167.5 (C=O), 141.7 (py-C3), 139.8 (py-C2), 138.2 (py-C6), 123.0 (py-C4), 121.8 (py-C5), 61.5 (O-CH₃), 33.3 (N-CH₃); NMR assignments are based on HMBC and HSQC experiments; ESI-HRMS: calcd for $[M + H]^+ C_8H_{12}N_3O_2$: 182.0924; obsd 182.0931.

3-Aminoisonicotinaldehyde (13). A solution of Weinreb amide 12 (138 mg, 0.759 mmol) in dry THF (2 mL) was added dropwise over 15 min to a solution of $LiAlH₄$ (86.4 mg, 2.28 mmol) in dry THF (3 mL) at -15° C. The reaction was stirred for 1.5 h at -15° C and quenched by pouring the mixture into a phosphate buffer solution $(1 M, pH 7.5, 30 mL)$ at 0°C. The aqueous solution was extracted with ether (3 \times 20 mL), dried over $Na₂SO₄$ and concentrated to give the title compound 13 as a yellow solid, 83 mg (90%) , pure by ¹H NMR; ¹H NMR (400 MHz, CDCl₃): δ_H 9.97 (s, 1H, -CHO), 8.24 (s, 1H, py-H2), 8.07 (d, $J = 5.2$ Hz, 1H, py-H6), 7.33 (d, $J = 5.2$ Hz, 1H, py-H5), 6.00 (br s, 2H, NH₂).

General procedure for the formation of the 1,7-naphthyridines 16–18. 3-Aminoisonicotinaldehyde (13) in dry THF (1 mL) was added to a solution NaH in dry THF (1 mL) at 0° C. The appropriate ketone 4, 14, or 15 in dry THF (2 mL) was added dropwise over 10 min. The reaction was stirred for 15 min at 0° C before it was allowed to heat to room temperature and stirred for 2 h. Water (15 mL) was added and the mixture was extracted with EtOAc $(3 \times 20 \text{ mL})$. The combined organic extracts were dried over $Na₂SO₄$ and the solvent was removed under reduced pressure. The products 16–18 were isolated after purification by flash column chromatography.

2-(Pyridin-4-yl)-1,7-naphthyridine (16). The title compound was prepared form aminoaldehyde 13 (34.0 mg, 0.278 mmol), ketone 4 (37.0 mg, 0.305 mmol) and NaH (15.0 mg, 0.625 mmol). After flash column chromatography (gradient: 5–10% MeOH/CH₂Cl₂), product 16 [19] was isolated as a white solid, 41 mg (71%), mp 157–158 °C, pure by NMR; R_f 0.45 (10%) MeOH/CH₂Cl₂); IR (KBr): 3039, 1596, 1489, 1420, 948, 828, 788, 705, 682 cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ_H 9.61 (s, 1H, H8), 8.82 (dd, $J = 4.8$, 1.6 Hz, 2H, py'-H2, -H6), 8.66 (d, J $=$ 5.6 Hz, 1H, H6), 8.30 (d, $J =$ 8.4 Hz, 1H, H4), 8.11 (d, $J =$ 8.4 Hz, 1H, H3), 8.08 (dd, $J = 4.8$, 1.6 Hz, 2H, py'-H3, -H5), 7.70 (d, $J = 5.6$ Hz, 1H, H5); ¹³C NMR (100 MHz, CDCl₃): δ_C 156.3 (C2), 154.9 (C8), 150.9 (py'-C2, -C6), 145.7 (py'-C4), 144.4 (C6), 143.4 (C8a), 136.2 (C4), 130.7 (C4a), 122.6 (C3), 121.6 (py'-C3, -C5), 119.7 (C5); NMR assignments are based on HMBC and HSQC experiments; ESI-HRMS: calcd for $[M +]$ $[H]^+$ C₁₃H₁₀N₃: 208.0869; obsd 208.0880; calcd for $[M + Na]^+$ $C_{13}H_9N_3Na$: 230.0689; obsd 230.0691.

2-(Naphthalen-1-yl)-1,7-naphthyridine (17). The title compound was prepared form aminoaldehyde 13 (29.0 mg, 0.237 mmol), ketone 14 (42.0 mg, 0.247 mmol), and NaH (15.0 mg, 0.625 mmol). After flash column chromatography [EtOAc/pentane $(1:1)$], product 17 was isolated as a white solid, 19 mg (31%), mp 116–117°C, pure by NMR; R_f 0.27 [EtOAc/pentane] (1:1)]; IR: 3045, 1598, 1496, 1409, 1241, 942, 855, 803, 789, 773 cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ_H 9.64 (s, 1H, H8), 8.69 (d, $J = 5.6$ Hz, 1H, H6), 8.28 (d, $J = 8.4$ Hz, 1H, H4), 8.12 (m, 1H, Np'-H), 7.97 (m, 2H, Np'-H), 7.92 (d, $J = 8.4$ Hz, 1H, H3), 7.74 (d, $J = 5.6$ Hz, 1H, H5), 7.72 (m, 1H, Np'-H), 7.62 (m, 1H, Np'-H) 7.52 (m, 2H, Np'-H); ¹³C NMR (100 MHz, CDCl₃): δ_C 161.4 (C2), 154.7 (C8), 144.1 (C6), 143.4 (C8a), 138.0 (Np'-Cq), 135.1 (C4), 134.2 (Np'-Cq), 131.2 (Np'-Cq), 130.1 (C4a), 129.9 (Np'-C), 128.8 (Np'-C), 128.3 (Np'-C), 127.5 (C3), 127.1 (Np'-C), 126.4 (Np'-C), 125.6 (Np'-C), 125.5 (Np'-C), 119.9 (C5); NMR assignments are based on HMBC and HSQC experiments; ESI-HRMS: calcd for $[M + H]^+$ C₁₈H₁₃N₂: 257.1073; obsd 257.1082.

3-Phenyl-2-(pyridin-4-yl)-1,7-naphthyridine (18). The title compound was prepared form aminoaldehyde 13 (47.0 mg, 0.385 mmol), ketone 15 (82 mg, 0.416 mmol) and NaH (19.0 mg, 0.792 mmol). After flash column chromatography (gradient: $5-10\%$ MeOH/CH₂Cl₂), product 18 was isolated as an orange solid, 30 mg (28%), mp 131-132°C, pure by NMR; R_f 0.16 (5%)

MeOH/CH₂Cl₂); IR (KBr): 3030, 1598, 1586, 1408, 970, 909, 831, 819, 767, 701 cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ_H 9.61 (s, 1H, H8), 8.68 (d, $J = 5.6$ Hz, 1H, H6), 8.57 (d, $J = 6.0$ Hz, 2H, py'-H2, -H6), 8.21 (s, 1H, H4), 7.72 (d, $J = 5.6$ Hz, 1H, H5), 7.36 (m, 5H, py'-H3, -H5, Ph-H3, -H4, -H5), 7.25 (m, 2H, Ph-H2, -H6); ¹³C NMR (100 MHz, CDCl₃): δ_c 157.6 (C2), 154.6 (C8), 149.9 (py'-C2, -C6), 147.3 (py'-C4), 144.7 (C6), 142.4 (C8a), 138.7 (C3), 138.3 (Ph-C1), 136.6 (C4), 130.5 (C4a), 129.7 (Ph-C2, -C6), 128.9 (Ph-C3, -C5), 128.6 (Ph-C4) 124.5 (py'-C3, -C5), 119.7 (C5); NMR assignments are based on HMBC and HSQC experiments; ESI-HRMS: calcd for $[M +]$ $[H]^+ C_{19}H_{14}N_3$: 284.1182; obsd 284.1185; calcd for $[M + Na]^+$ C19H13N3Na: 307.1032; obsd 307.1032.

REFERENCES AND NOTES

[1] Marco-Contelles, J.; Perez-Mayoral, E.; Samadi, A.; Carreiras, M. C.; Soriano, E. Chem Rev 2009, 109, 2652.

[2] Cheng, C.-C.; Yan, S.-J. Org React 1982, 28, 37.

[3] Bose, D. S.; Idrees, M.; Jakka, N. M.; Rao, J. V. J Comb Chem 2010, 12, 100.

[4] Mohammadi, A. A.; Azizian, J.; Hadadzahmatkesh, A.; Asghariganjeh, M. R. Heterocycles 2008, 75, 947.

[5] Rahman, A. F. M. M.; Kwon, Y.; Jahng, Y. Heterocycles 2005, 65, 2777.

[6] Yang, D.-Q.; Zhong, G.-F.; Guo, W.; Zeng, H.-P.; Cao, L.; Liang, J.-C. Hecheng Huaxue 2005, 13, 381.

[7] Ubeda, J. I.; Villacampa, M.; Avendano, C. Synlett 1997, 285.

[8] Ravichandran, S.; Subramani, K.; Arunkumar, R. Int J Chem Sci 2009, 7, 993.

[9] Mogilaiah, K.; Sakram, B. Indian J Chem 2006, 45B, 2749.

[10] Mogilaiah, K.; Rani, J. U. Indian J Chem 2006, 45B, 1051.

[11] Zhichkin, P.; Cillo Beer, C. M.; Rennells, W. M.; Fairfax, D. J. Synlett 2006, 379.

[12] Mogilaiah, K.; Prashanthi, M.; Kavitha, S. Indian J Chem 2006, 45B, 302.

[13] Mogilaiah, K.; Sudhakar, G. R. Indian J Chem 2003, 42B, 1170.

[14] Mogilaiah, K.; Reddy, N. V. Indian J Chem 2002, 41B, 215.

[15] Litvinov, V. P. Adv Heterocycl Chem 2006, 91, 189.

[16] Phuan, P.-W.; Kozlowski, M. C. Sci Synth 2005, 15, 947.

[17] Litvinov, V. P.; Roman, S. V.; Dyachenko, V. D. Russ Chem Rev 2001, 70, 299.

[18] Turner, J. A. J Org Chem 1983, 48, 3401.

[19] Decormeille, A.; Guignant, F.; Queguiner, G.; Pastour, P. J Heterocycl Chem 1976, 13, 387.

[20] Wissner, A.; Hamann, P. R.; Nilakantan, R.; Greenberger, L. M.; Ye, F.; Rapuano, T. A.; Loganzo, F. Bioorg Med Chem Lett 2004, 14, 1411.

[21] Kaila, N.; Green, N.; Li, H.-Q.; Hu, Y.; Janz, K.; Gavrin, L. K.; Thomason, J.; Tam, S.; Powell, D.; Cuozzo, J. Bioorg Med Chem 2007, 15, 6425.

[22] Bakke, J. M.; Hegbom, I.; Øvreeide, K.; Aaby, K. Acta Chem Scand 1994, 48, 1001.

[23] Bakke, J. M.; Ranes, E. Synthesis 1997, 281.

[24] Bakke, J. M.; Riha, J. J Heterocycl Chem 2001, 38, 99.

[25] Shimizu, T.; Osako, K.; Nakata, T. Tetrahedron Lett 1997, 38, 2685.

[26] Sheldrake, P. W. Synth Commun 1993, 23, 1967.